





Evaluation of Regional-Scale Model Parameters in the Prediction of Isoprene Epoxydiol (IEPOX)-Derived Secondary Organic Aerosols (SOA) Generated during Laboratory Chamber Experiments

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Isoprene epoxydiol (IEPOX) secondary organic aerosols (SOA) are formed at rural/urban interfaces



Hu et al. (2015) Atmos. Chem. Phys.





IEPOX SOA chemical processes



Modeling challenges

Critical physiochemical properties:

- Phase state/separation
- Acid-catalyzed reactive uptake
- Kinetics

Environmental conditions:

- Relative humidity
- IEPOX:Sulfate ratios



Regional-scale models with phase separation underpredict IEPOX SOA



Farrell et al. Masters Thesis (2021) University of North Carolina at Chapel Hill Farrell et al. (2023) In prep Schmedding et al. (2020) Atmos. Chem. and Physics



Remaining model uncertainties...

- Reactive uptake dependent on complex physiochemical properties and environmental conditions
- Organics not directly accounted for in regional-scale thermodynamic models for acidity
- Single heterogeneous rate constant controls total IEPOX SOA formed which is then fractioned into 2-MT, 2-MTS, and dimers







1. What are the critical parameters in IEPOX SOA multiphase chemistry?

2. How can IEPOX SOA and other multiphase reactions be modeled more accurately and efficiently in regional-scale models?

3. What are the implications of this chemistry on rural/urban interaction in a changing environment?

Riva & Chen et al. (2019) Environ. Sci. & Tech.







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9 chamber experiments at variable IEPOX:Sulfate ratios in humid (50% RH) and dark conditions



Chen Thesis (2021) UNC



Modeling experiments with CMAQ aerosol modules

Community Multiscale Air Quality (CMAQ) Model

- Utilized and developed by the U.S. Environmental Protection Agency (EPA)
- Uses FORTRAN and is run on computing cluster
- Computationally expensive



3-D



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Chamber modeling with CMAQ modules in MATLAB

- CMAQ 5.3 modules translated to MATLAB by Jaime Green
- Can simulate CMAQ treatment of heterogeneous chemistry
- Computationally efficient for physiochemical parameter analysis



3-D



Modified CMAQ explicit multiphase chemistry model includes...

- Phase state/separation prediction*
 - Phase separated or homogeneous
 - Liquid or semi-solid organic shell
- Organic shell viscosity prediction
- Incorporates new scientific research on kinetics and reactive uptake

*Schmedding et al. (2020) Atmos. Chem. and Physics













CMAQ homogeneous model underpredicted IEPOX SOA and phase separation model worsened predictions



Conversion of inorganic to organic sulfate reduces predicted acidity when organics are not accounted for in thermodynamic models



Homogeneous -----

Phase separated - - - -

Heterogeneous rate constant sensitivity

- Constant parameters at time = 50 minutes
- Heterogeneous rate constant as a function of high and low literature values
- Flat line = insensitive



 $IEPOX_{gas} \rightarrow IEPOXSOA_{aero}$

$$k_{het} = \frac{SA}{\frac{r_p}{D_g} + \frac{4}{\vartheta\gamma}}$$

Parameter	Average Percent Change
Surface area (cm ² cm ⁻³)	$1.00 imes 10^4$
Reactive uptake	894
Particle radius (cm)	-0.106
Gas-phase diffusivity (cm ² s ⁻¹)	0.023





Model prediction improved with increased organic shell diffusivity IEPOX:Sulf = 10.5



18

Utilizing homogeneous and phase state/separation reactive uptake for different reactions improved model performance





The most sensitive model parameters were:

- Mass accommodation
- Particle size (surface area, radius, shell thickness)
- Diffusivity through the organic shell

Phase state/separation reactive uptake model has little sensitivity to acidity changes

Increasing organic shell diffusivity resulted in improved model performance

Utilizing two separate coefficients for reactive uptake of IEPOX to acidified sulfate and pre-existing IEPOX SOA improved model performance

Modeling performance differed for **high and low IEPOX:Sulfate ratios.** As natural and anthropogenic emissions change, IEPOX:Sulfate is expected to increase.

Future work: Implement improved model in regional-scale

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Thank You!

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