

# Embedding Palladium into flame-made & leached SnO<sub>2</sub> drastically enhances Gas Sensing



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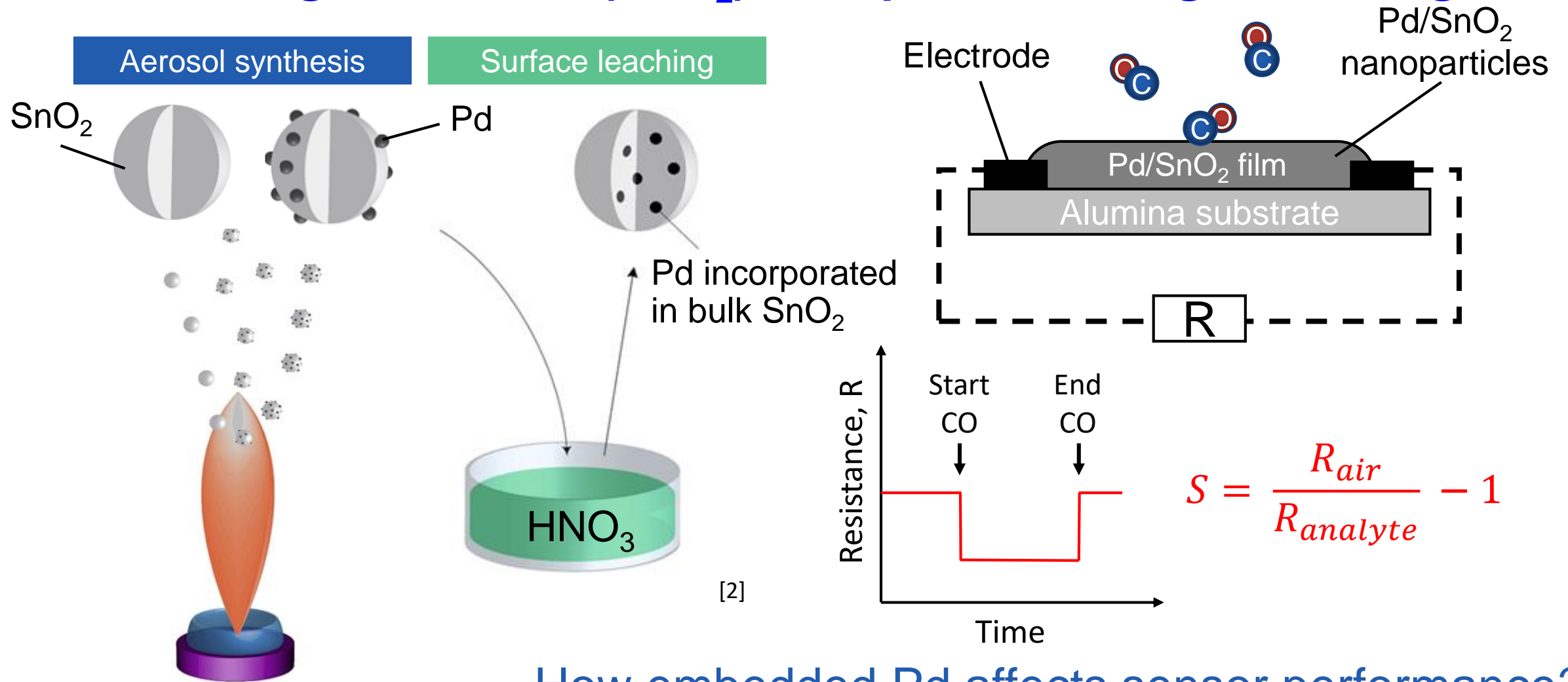
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Particle Technology Laboratory, Dept. of Mechanical & Process Eng'g, ETH Zurich, Switzerland



# Pd-containing tin dioxide (SnO<sub>2</sub>) nanoparticles for gas sensing



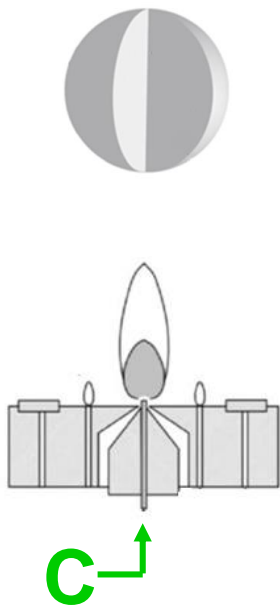
FSP uniquely embeds noble metals in ceramics<sup>1</sup>

How embedded Pd affects sensor performance?  
How can we control the surface & embedded Pd fractions?

1. van Vegten, N.; Maciejewski, M.; Krumeich, F.; Baiker, A., Structural properties, redox behaviour and methane combustion activity of differently supported flame-made Pd catalysts. *Appl Catal B-Environ* **2009**, *93*, 38-49.

2. Pineau NJ, Keller SD, Güntner AT, Pratsinis SE. Flame-made chemoresistive gas sensors and devices. *Microchim Acta*. 2020;187:96.

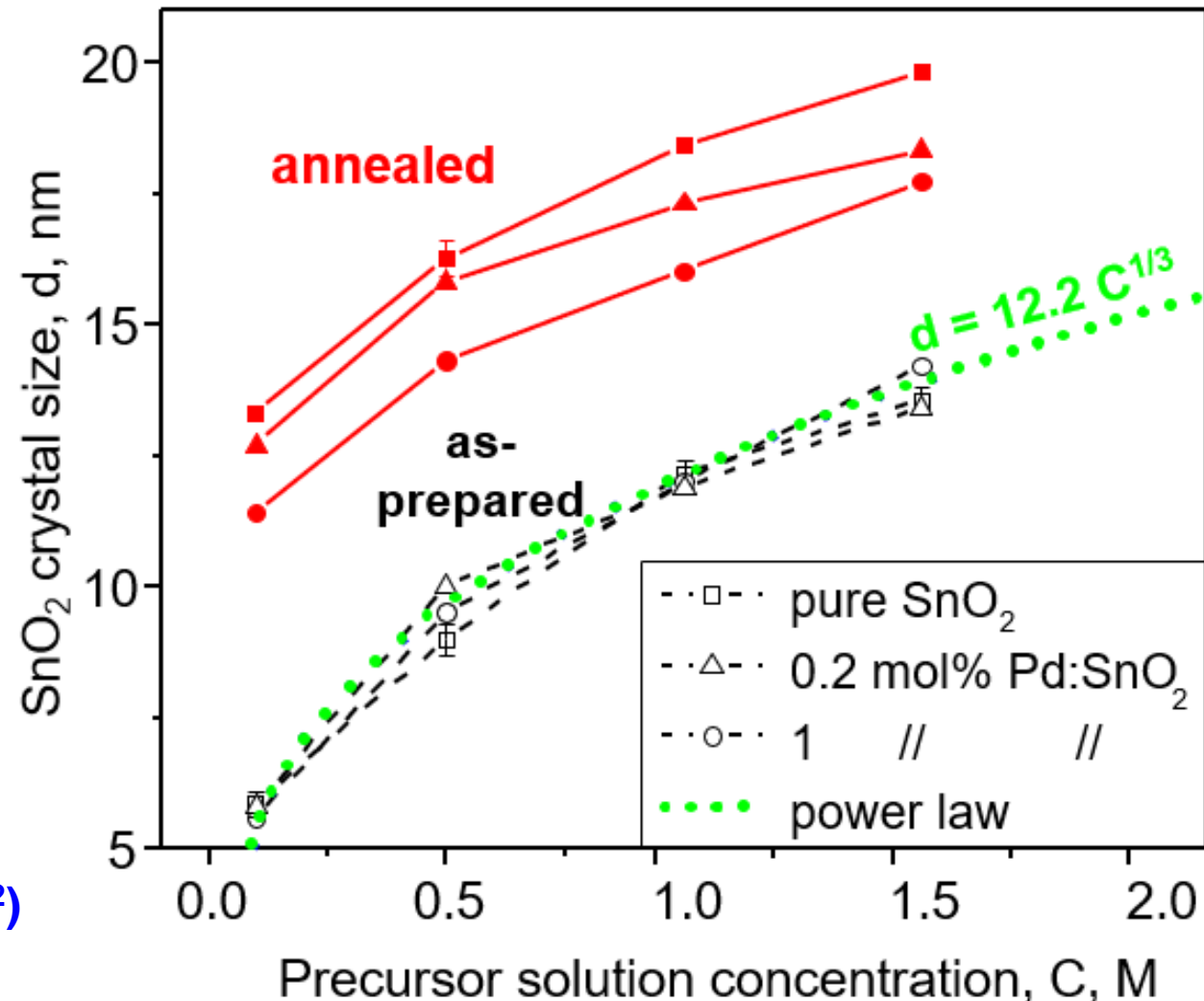
# Control of SnO<sub>2</sub> crystal size by Precursor solution concentration, C



Pd doesn't affect as-prepared SnO<sub>2</sub> particle size but decreases its annealed crystal size<sup>1</sup>

$d \sim C^{1/3} \rightarrow$  particle growth

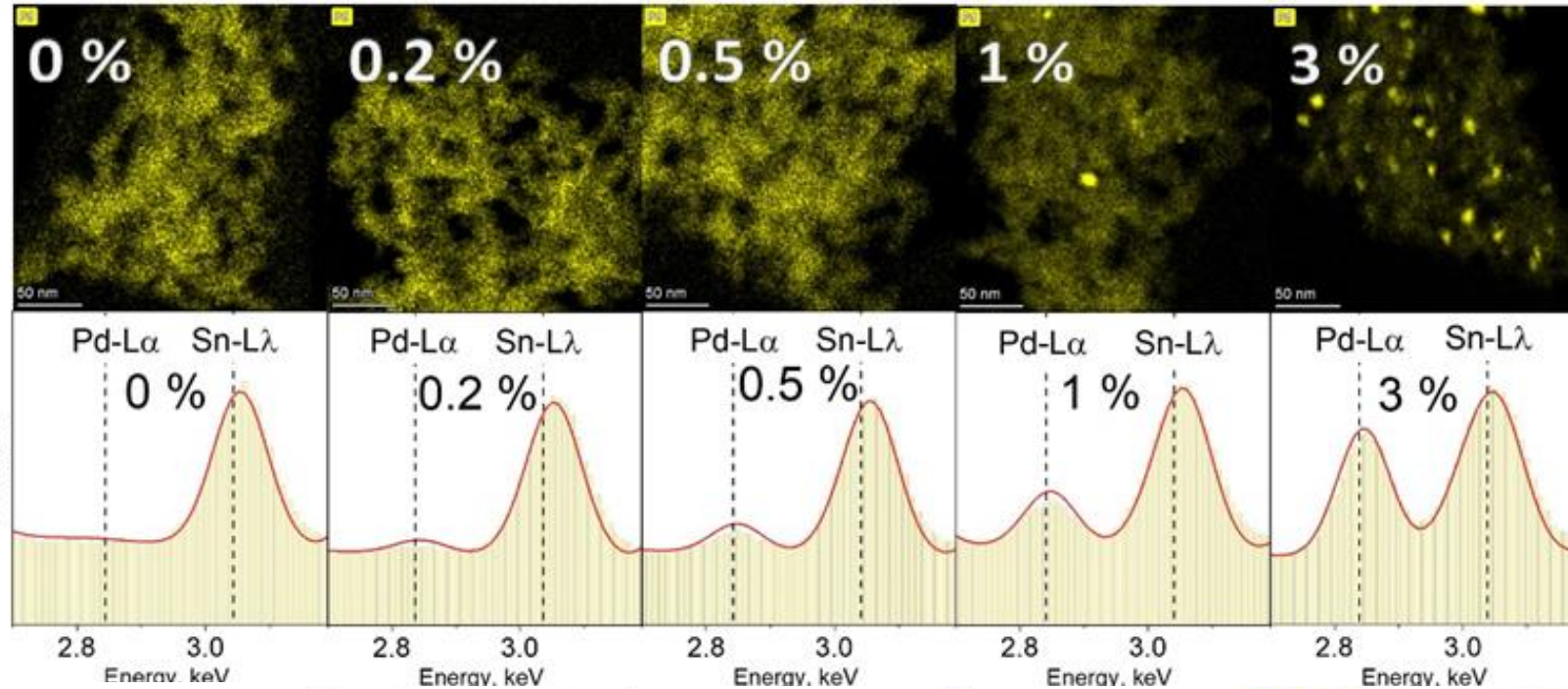
by droplet-to-particle formation (microexplosions<sup>2</sup>)  
rather than coagulation-sintering  $d \sim C^{2/5}$



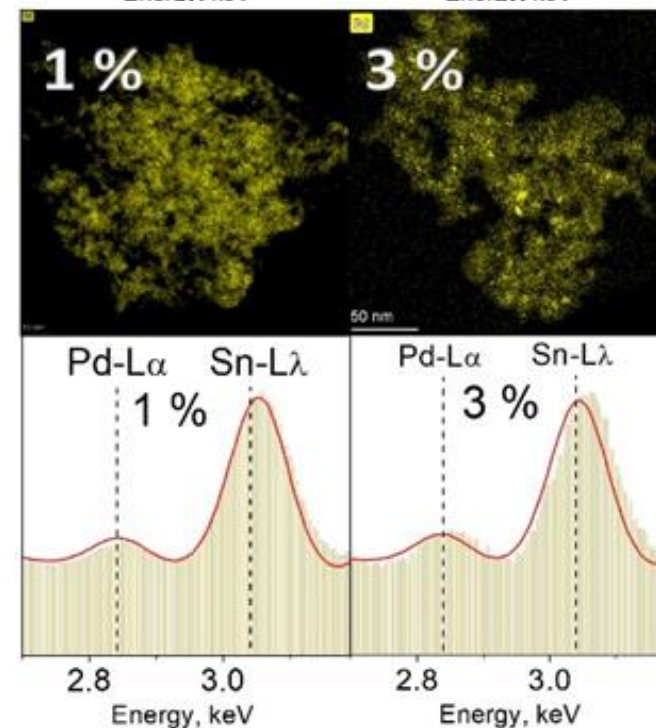
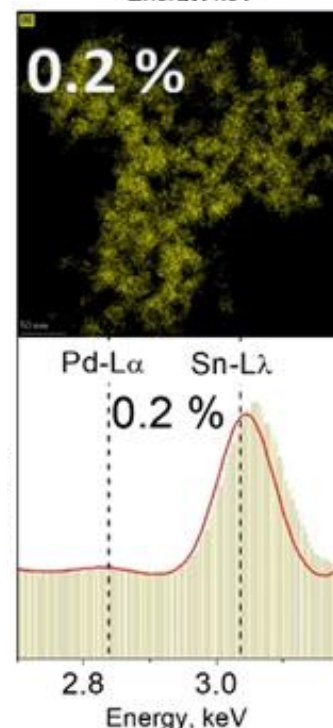
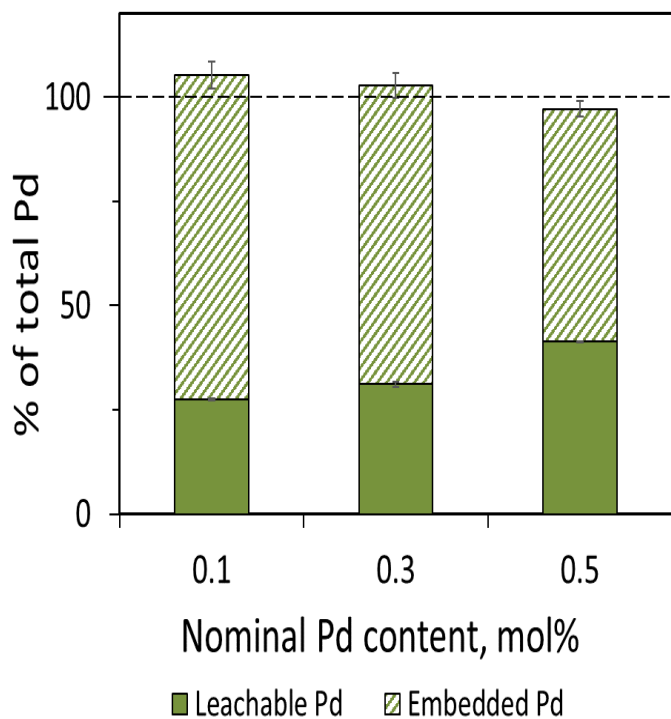
1. Pineau NJ, Keller SD, Güntner AT, SEP. Palladium embedded in SnO<sub>2</sub> enhances the sensitivity of flame-made chemoresistive gas sensors. *Microchim Acta*. 2020;187:96

2. C.D. Rosebrock, T. Wriedt, L. Mädler, K. Wegner, The role of microexplosions in flame spray synthesis for homogeneous nanopowders from low-cost metal precursors, *AIChE J.*, 2016, **62**, 381-391.

**Before leaching**

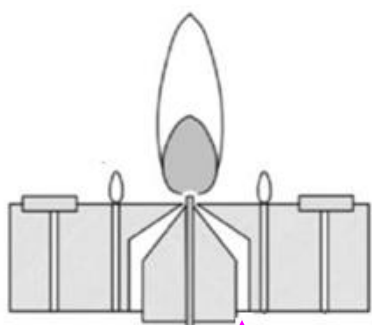
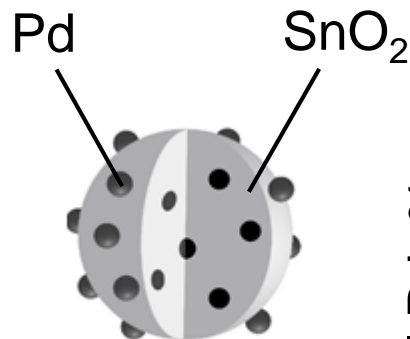


C=0.5 M, P/D=5/5  
ICP-OES of  
leachate and  
digested solution

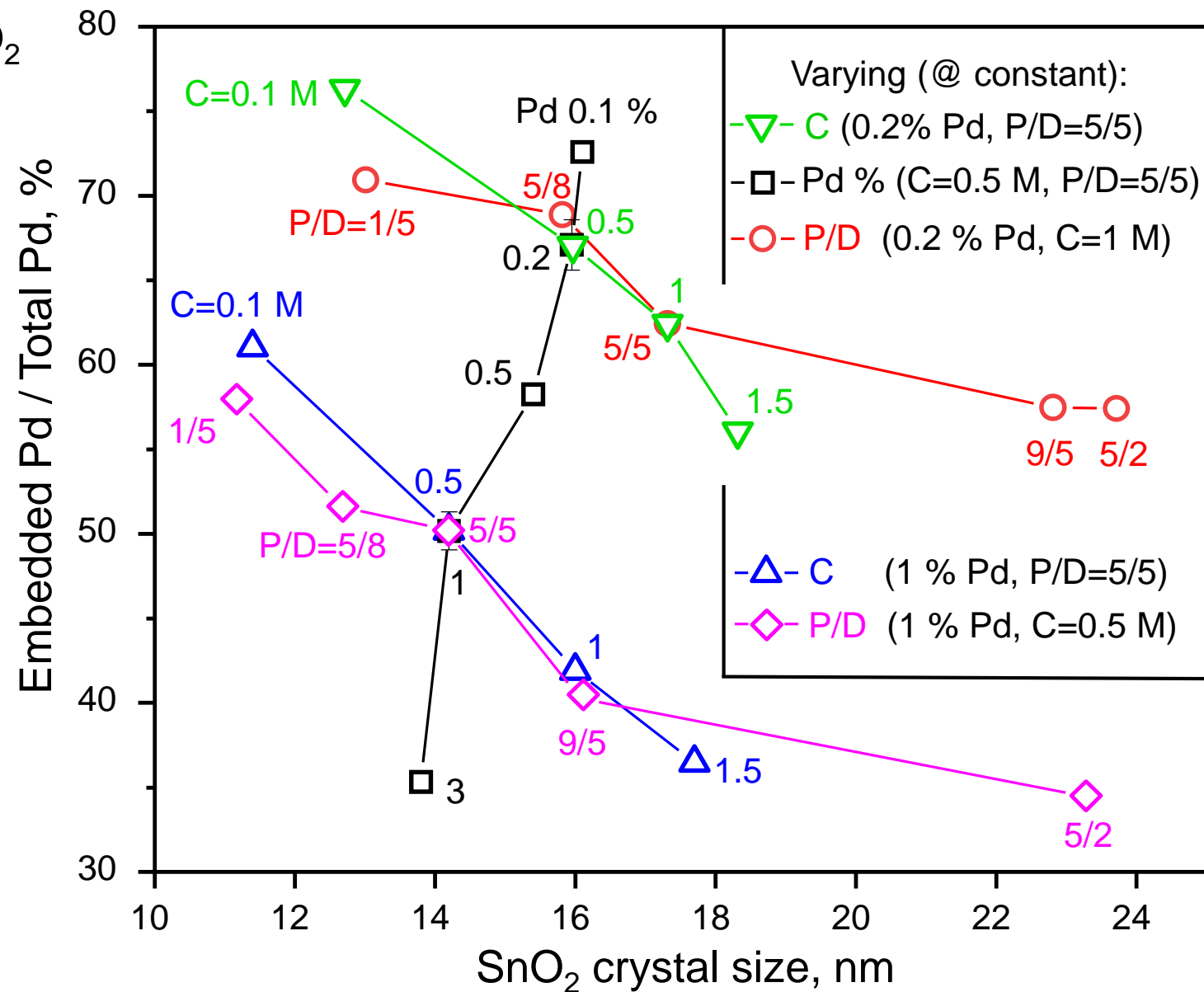


**After leaching**

# The Pd Fraction embedded in FSP-made Pd-containing SnO<sub>2</sub>



C  
P D  
Pd %

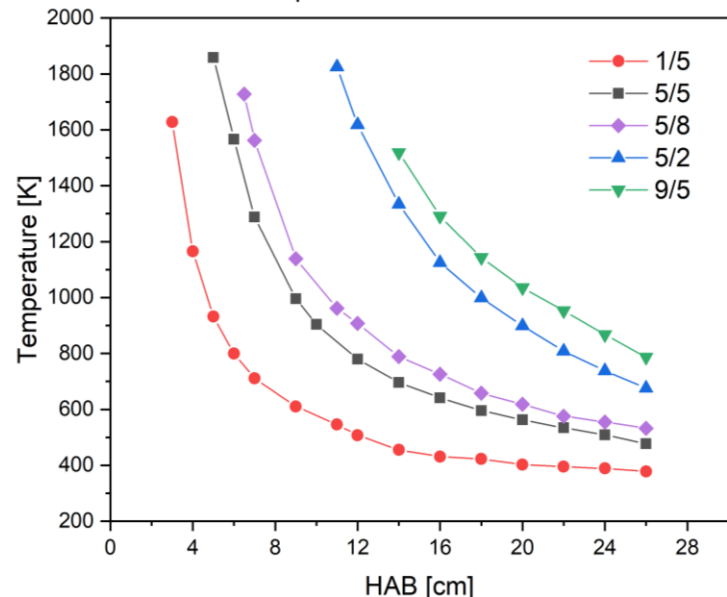
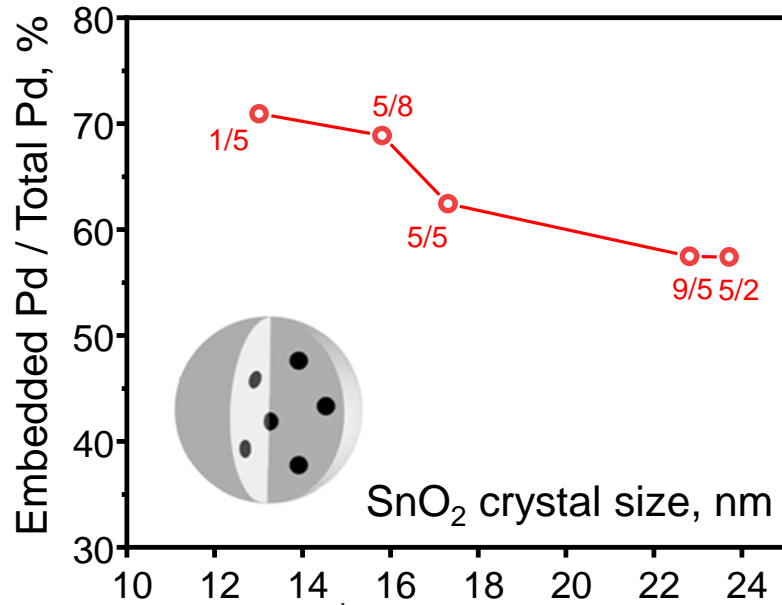


**30 – 80 %  
of Pd  
embedded  
into  
or  
20 – 70 %  
on the  
surface of  
SnO<sub>2</sub>!**

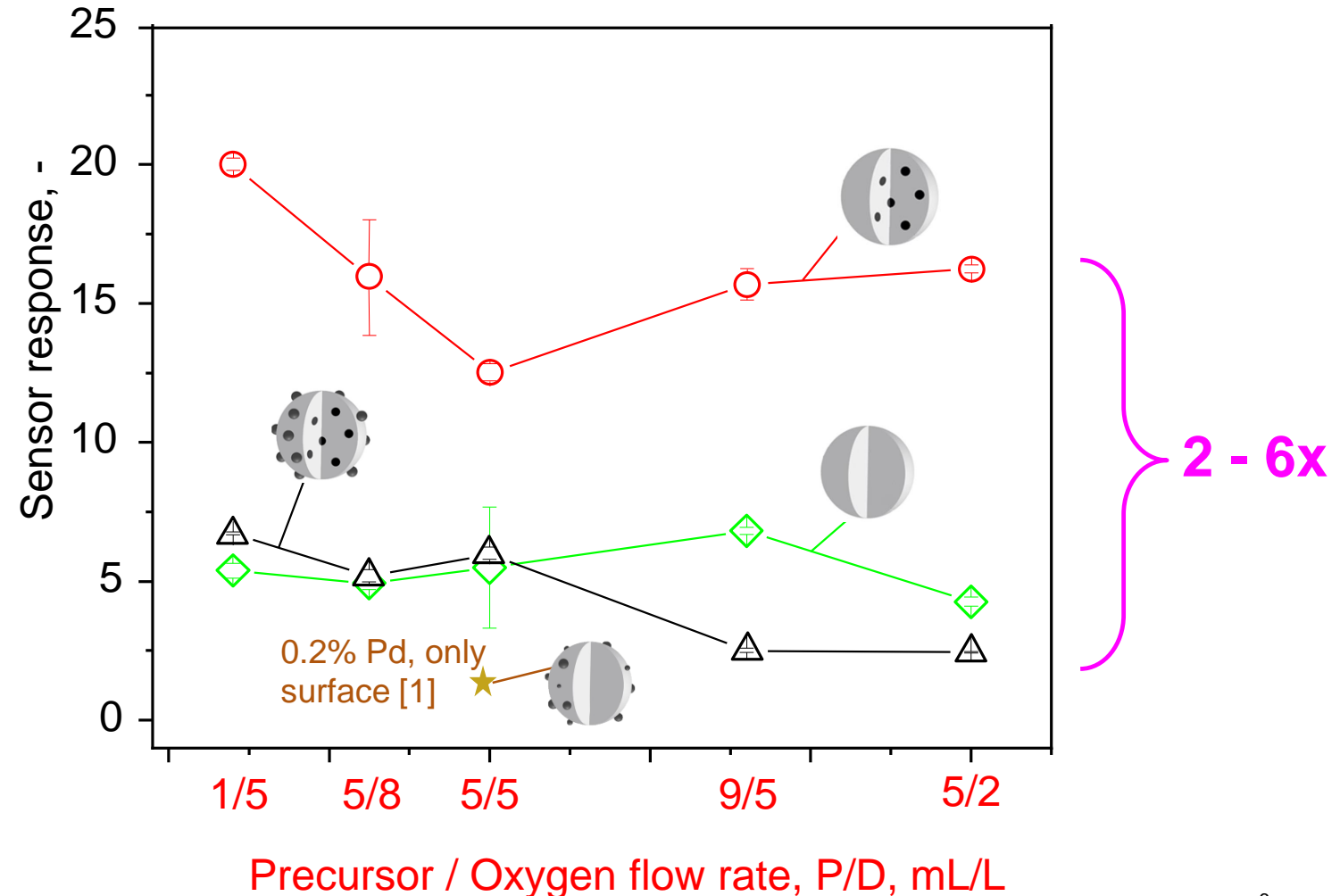
[1]

# Gas sensing performance

Varying the precursor / oxygen flow rate (P/D) @ 0.2 % Pd & C = 1 M



1 ppm Acetone (50% RH, 350 °C)

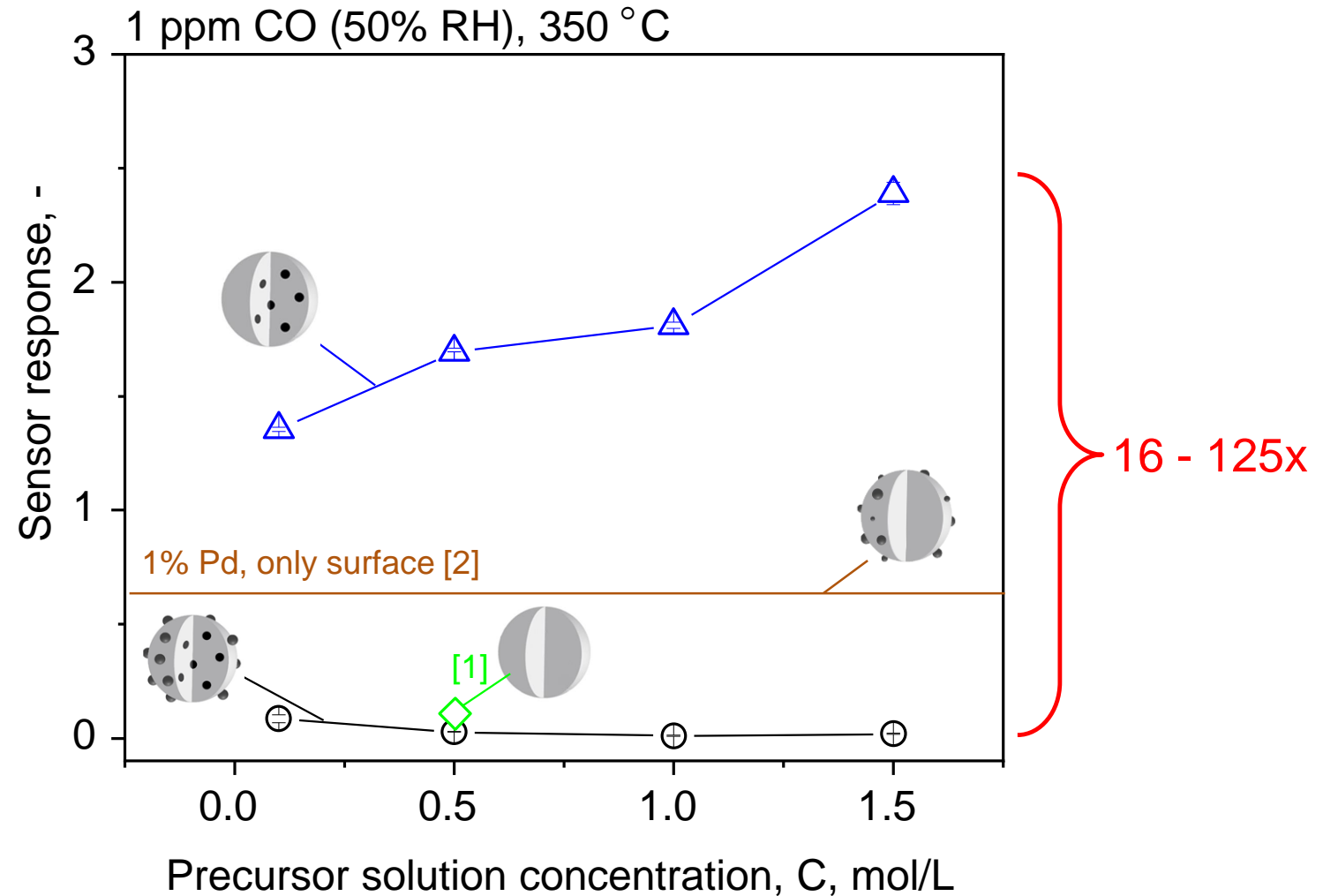
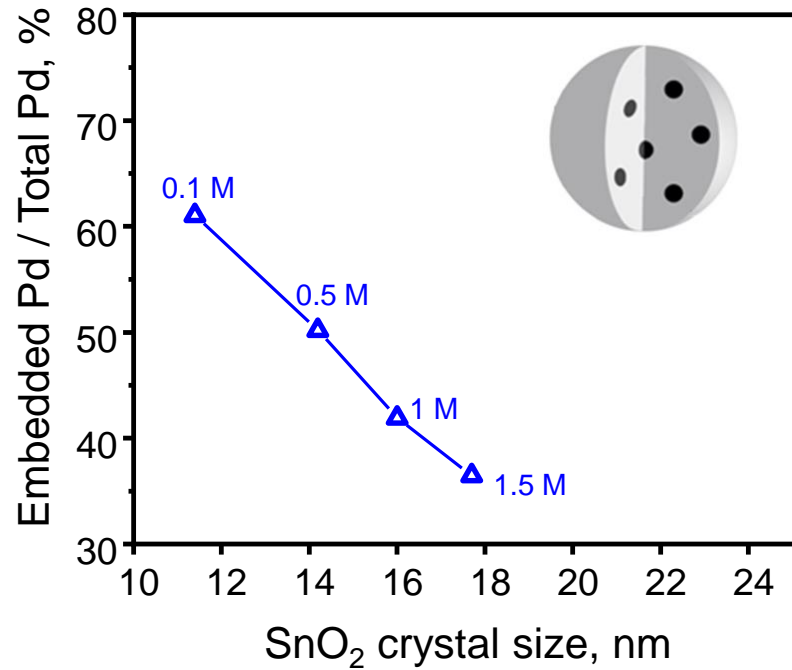


# Effect on precursor concentration on sensing performance

Varying the concentration (C)

1 % Pd

P/D = 5/5

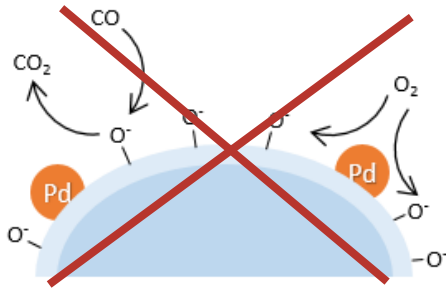


[1] Pineau NJ, Keller SD, Güntner AT, Pratsinis SE. Palladium embedded in SnO<sub>2</sub> enhances the sensitivity of flame-made chemoresistive gas sensors. *Microchim Acta*. 2020;187:96

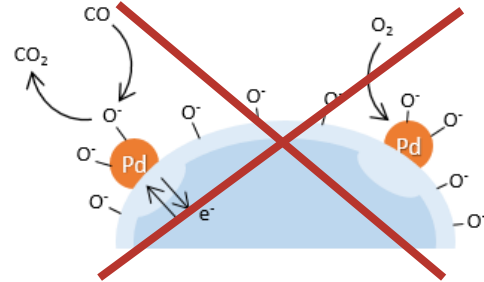
[2] Yuasa M et al. Nano-sized PdO loaded SnO<sub>2</sub> nanoparticles by reverse micelle method for highly sensitive CO gas sensor. *Sensor Actuat B*. 2009;136:99-104.

# Possible mechanisms of enhancement by embedded Pd

## Loading (Pd clusters):

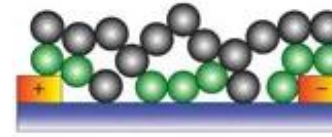


Spillover effect



Surface Fermi level control

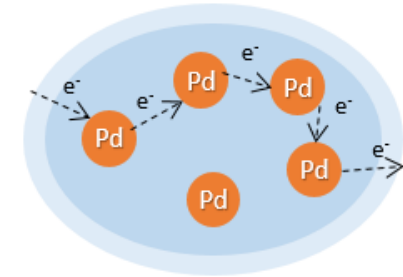
[1]



● Conductive particles

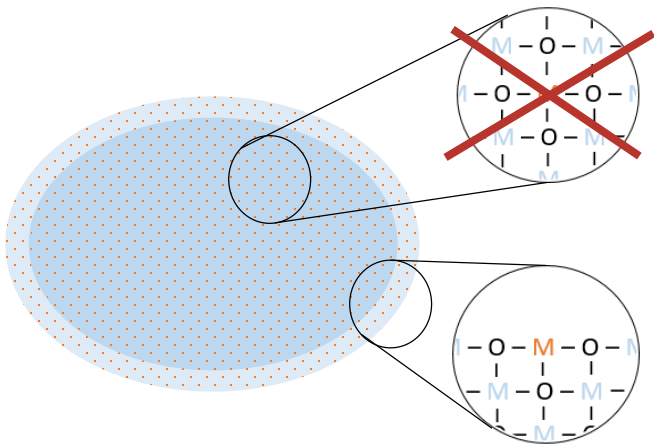
● Functional particles

[4]



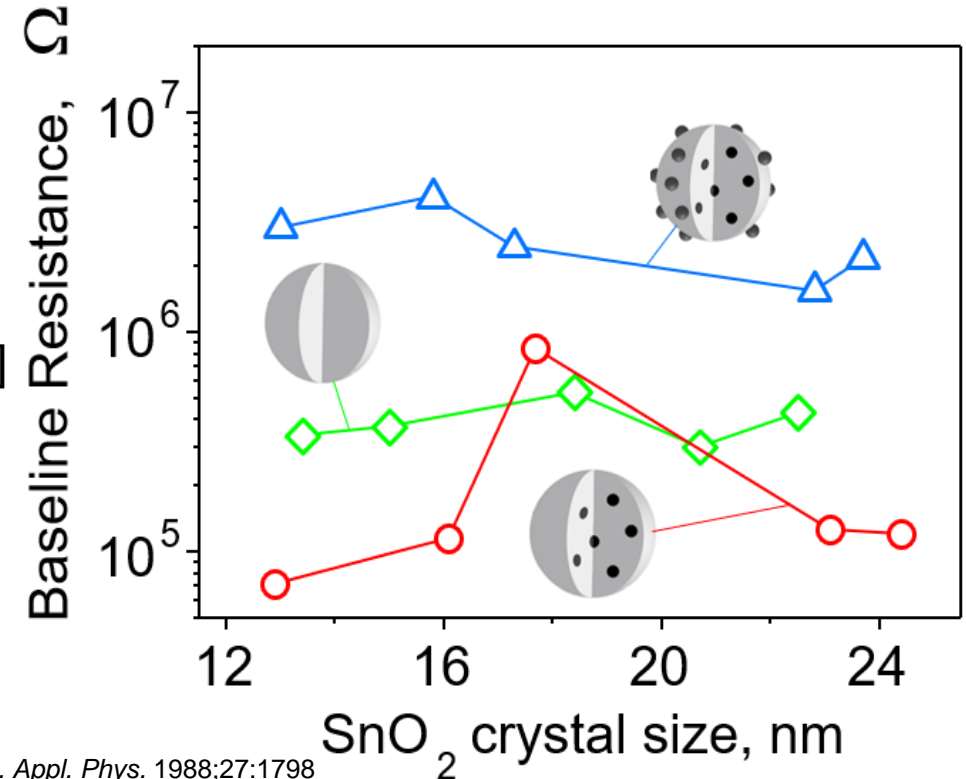
Conductive domains (nanoelectrodes)

## Doping (Pd ions):



New acceptor and donor states [2]

Direct activation of oxygen [3]



[1] Matsushima Sh, Teraoka Y, Miura N, Yamazoe N. Role of Additives on Alcohol Sensing by Semiconductor Gas Sensor. *Jpn. J. Appl. Phys.* 1988;27:1798

[2] Degler D, Pereira de Carvalho HW, Weimar U, Barsan N, Pham D, Mädler L, Grunwaldt J-D. Structure–function relationships of conventionally and flame made Pd-doped sensors studied by X-ray absorption spectroscopy and DC-resistance. *Sensor Actuat B-Chem.* 2015; 219:315-23.

[3] Koziej D, Hübner M, Barsan N, Weimar U, Sikora M, Grundwaldt J-D. Operando X-ray absorption spectroscopy studies on Pd-SnO<sub>2</sub> based sensors. *Phys. Chem. Chem. Phys.* 2009;11:8620-8625.

[4] Tricoli A, Pratsinis SE. Dispersed nanoelectrode devices. *Nature Nanotechnol.* 2010;5:54-60



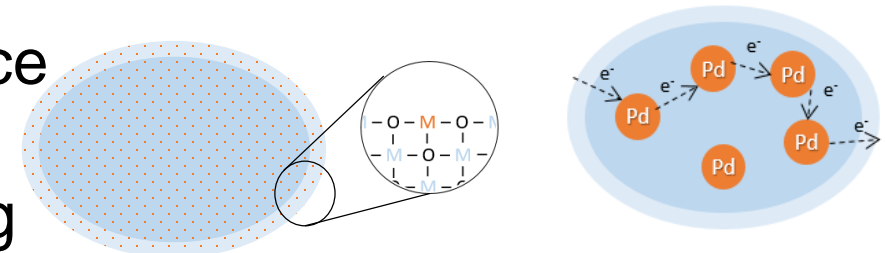
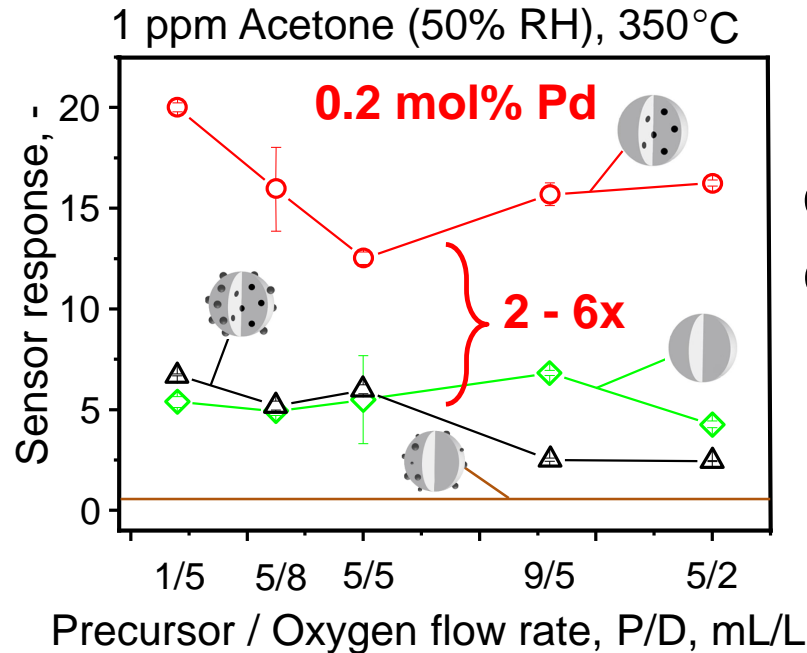
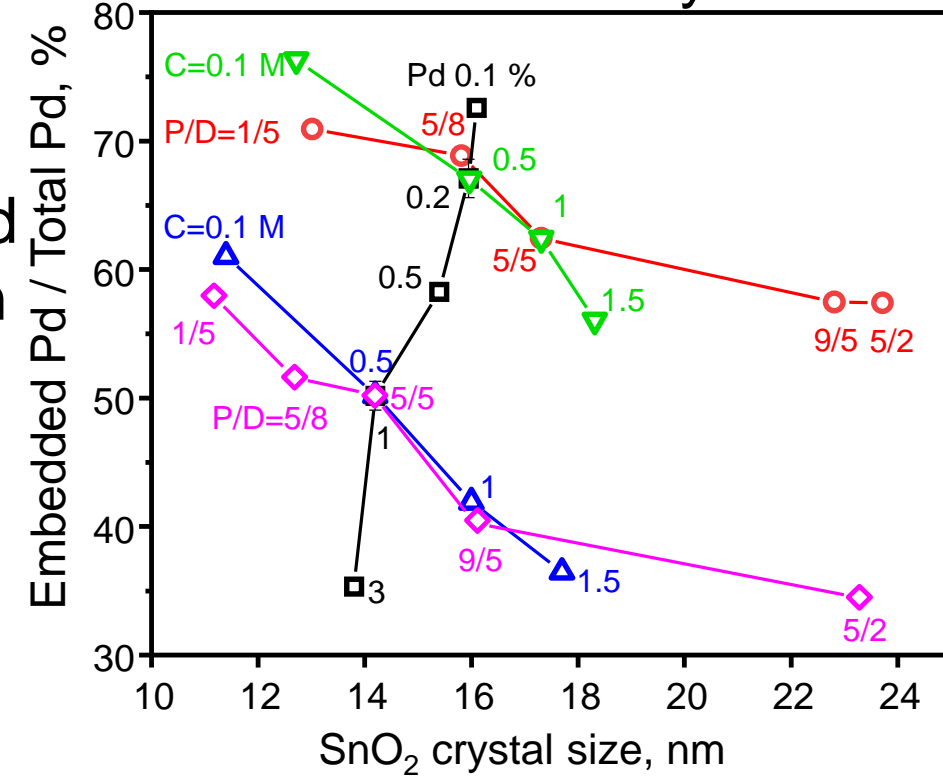
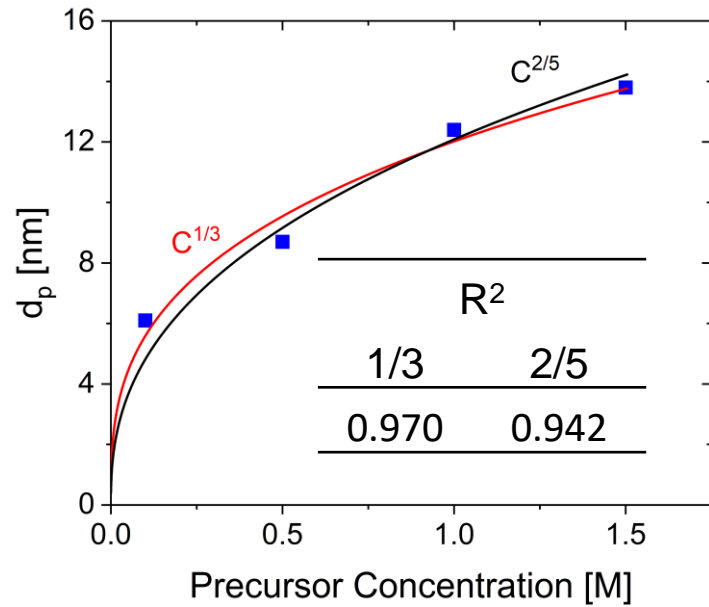
# Conclusions

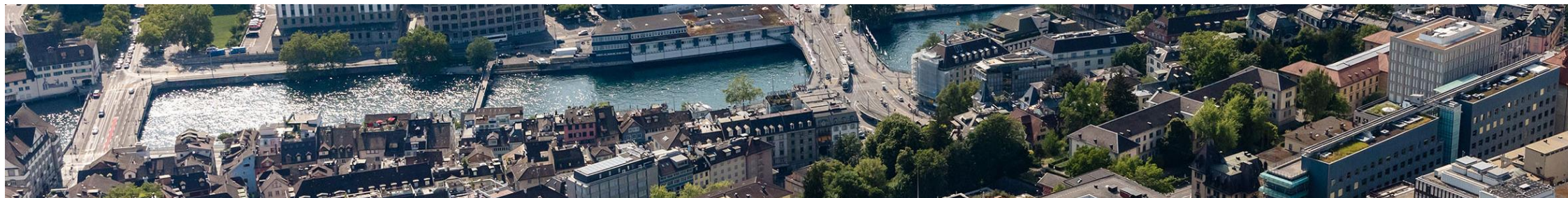
FSP-made  $\text{SnO}_2$  seems to form by droplet microexplosions followed by droplet-to-particle conversion rather than by coagulation-coalescence

Close control of Pd embedded fraction & crystal size

Drastic enhancement of gas sensing after embedding of Pd

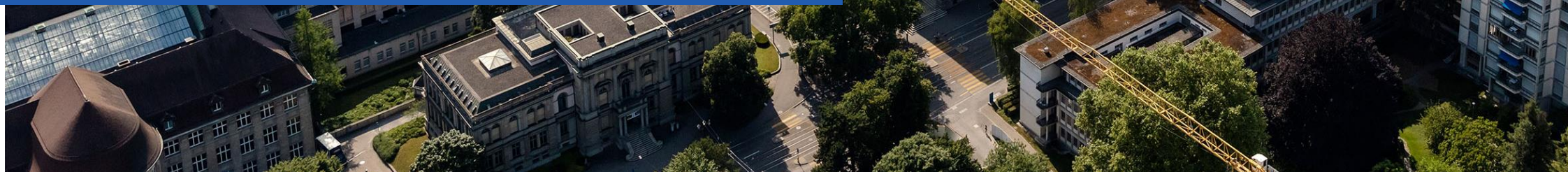
Bulk and/or surface mechanisms enhanced sensing





Thank you for your attention!

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Foundation (#159763 & 175754 and R'Equip 170729)



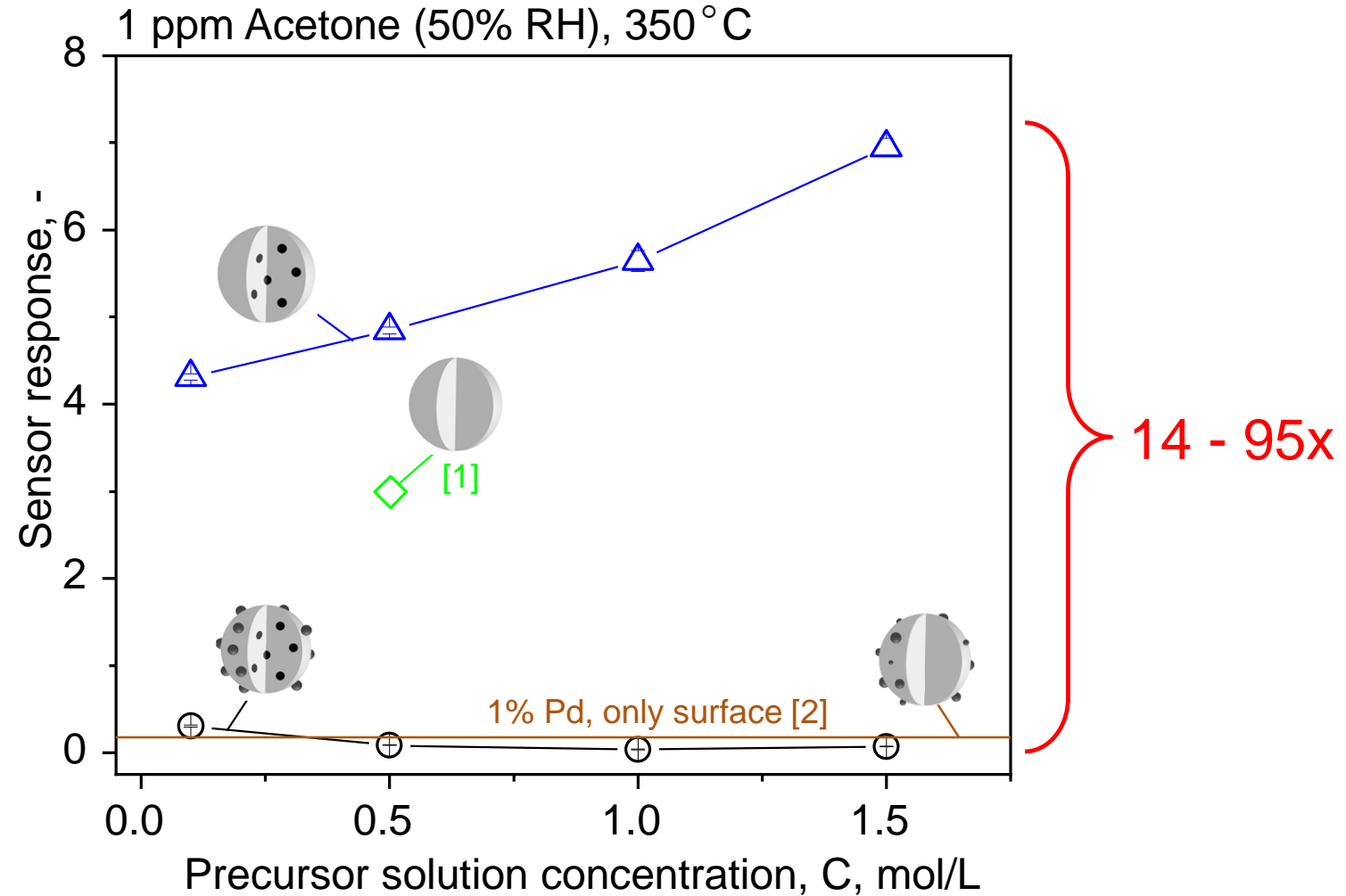
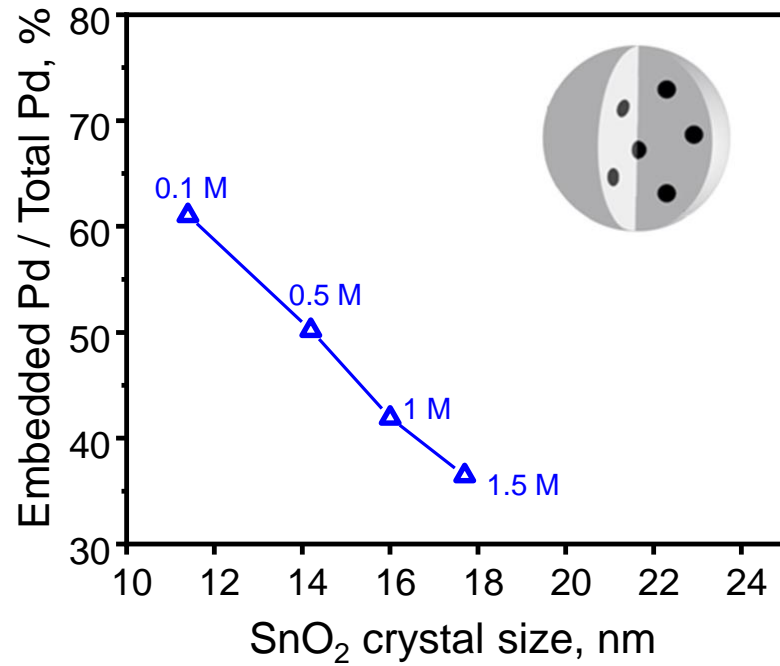
K. Jabłczyńska, A. Gogos, C.M.P. Kubsch, S.E. Pratsinis, Embedding Pd into SnO<sub>2</sub> drastically enhances gas sensing, *Nanoscale Advances*, **6**, 1259-1268 (2024).

# Effect on precursor concentration on sensing performance

Varying the concentration (C)

1 % Pd

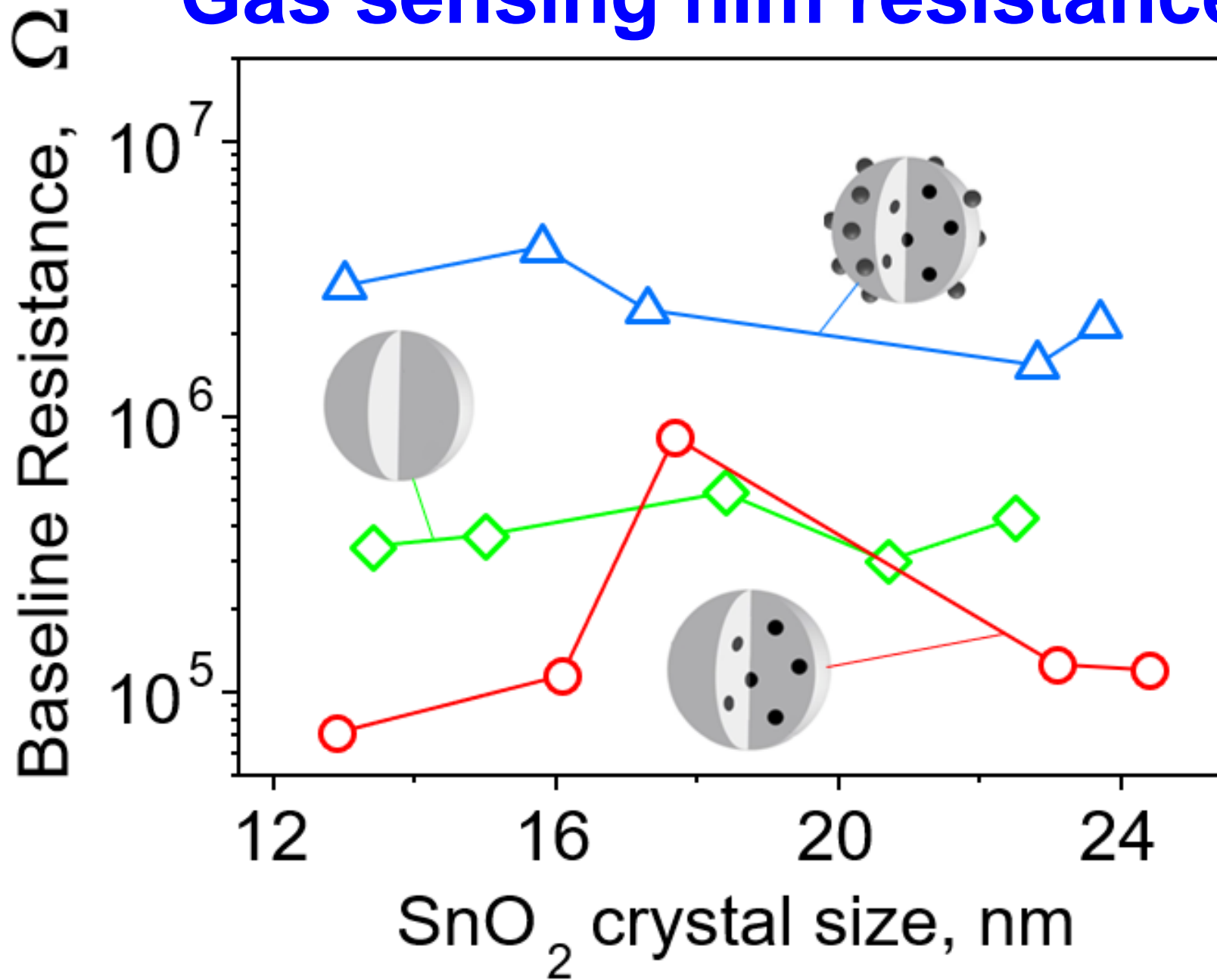
P/D = 5/5



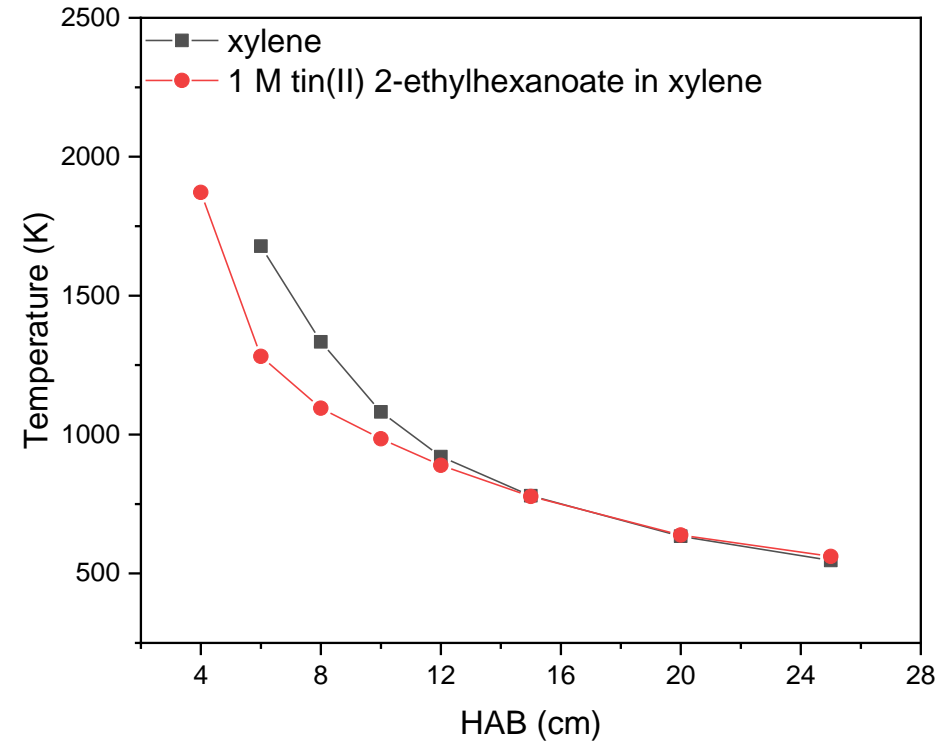
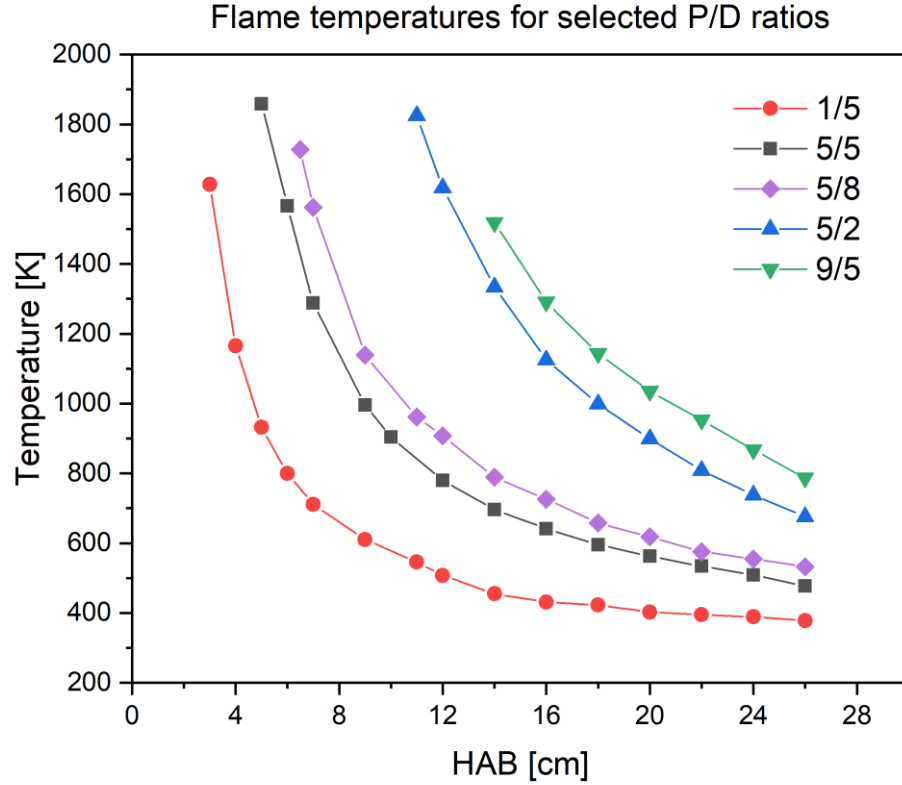
[1] Pineau NJ, Keller SD, Güntner AT, Pratsinis SE. Palladium embedded in SnO<sub>2</sub> enhances the sensitivity of flame-made chemoresistive gas sensors. *Microchim Acta*. 2020;187:96

[2] Gschwend P, Schenk F, Gogos A, Pratsinis SE, Acetone Sensing and Catalytic Conversion by Pd-Loaded SnO<sub>2</sub>. *Materials* 2021;14:5921.

# Gas sensing film resistance



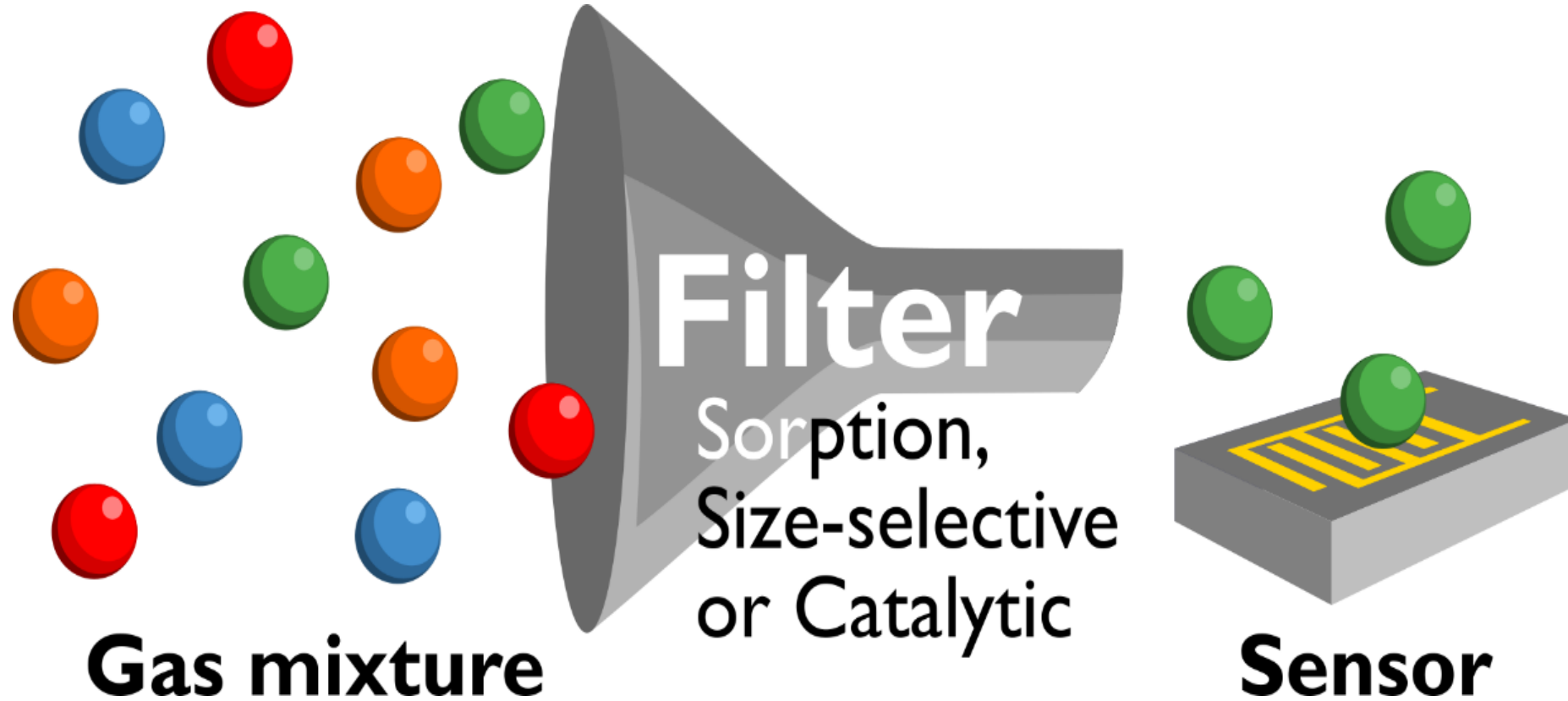
# Flame temperature profiles



## Boiling points of metallic and oxidized Pd and Sn

	$T_b$ [K]
Pd	3236
Sn	2875
SnO <sub>2</sub>	2173
SnO	1698
PdO	1023 (decomposition)

# Filter- or Concentrator-enhanced sensor selectivity

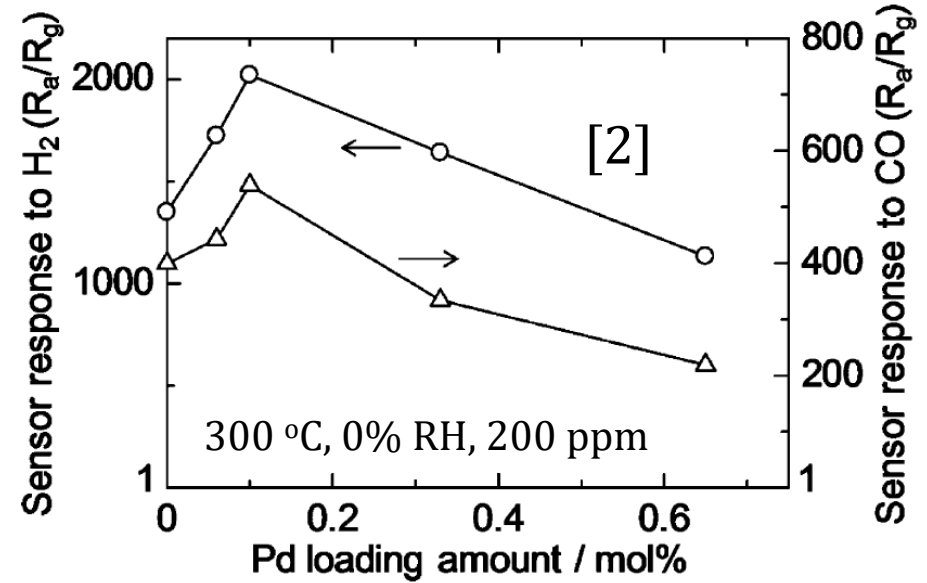
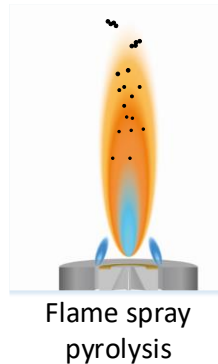
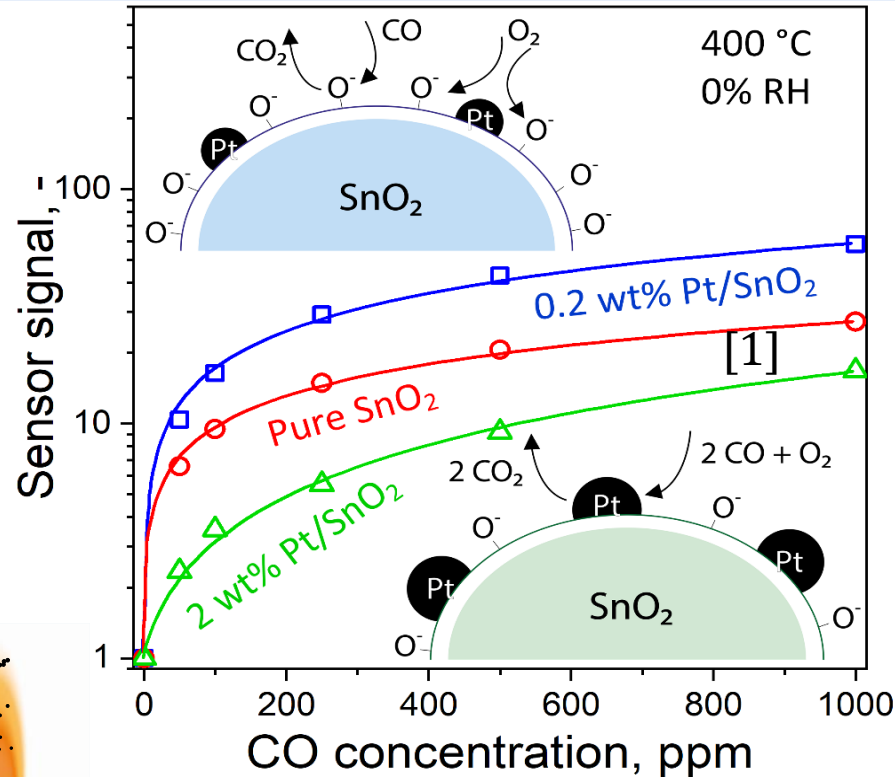


van den Broek J, Weber IC, Güntner AT, SEP. *Mater Horiz.* Highly selective gas sensing enabled by filters, **2021**;8:661-84

# Noble metal content in SnO<sub>2</sub> gas sensors

Korotcenkov G, Brinzari V, Boris Y, Ivanov M, Schwank J, Morante J. *Thin Solid Films*. 2003;436(1):119-126

spill-over” effect (i.e. *chemical sensitization*)



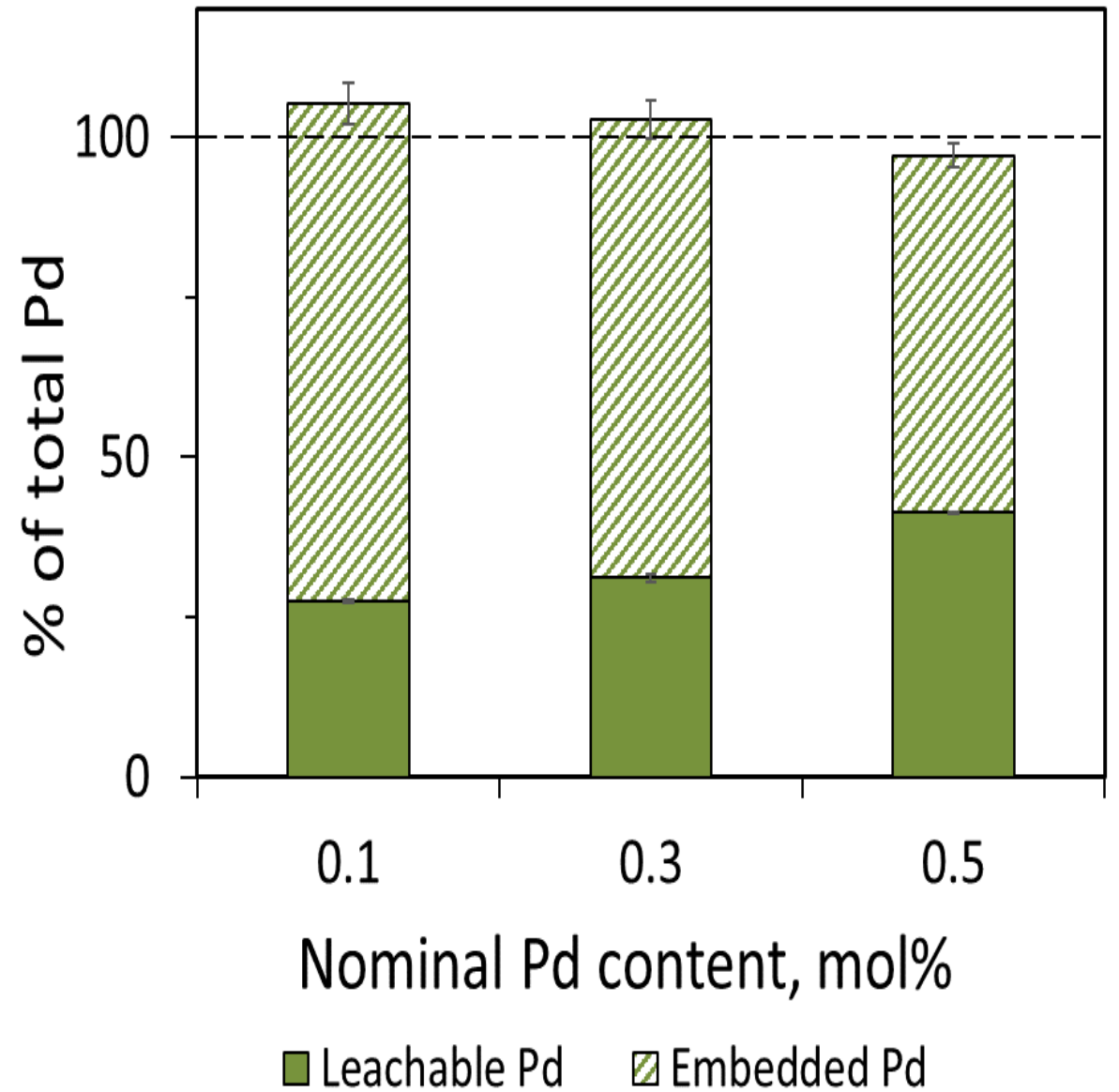
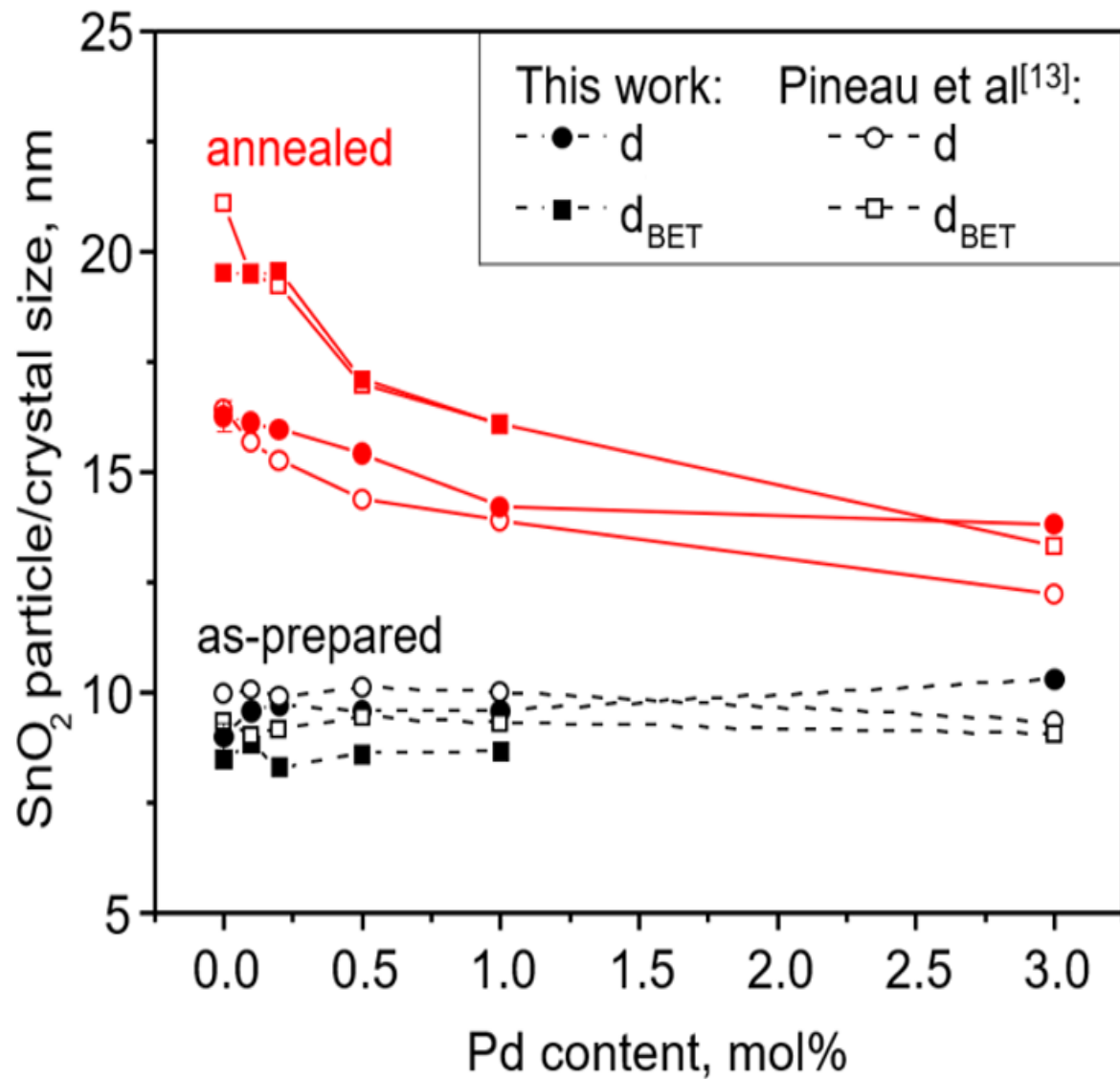
Wet made (precipitation) Pd on SnO<sub>2</sub>

- optimum at 0.15 mol% Pd
- sensor response enhanced with surface Pd

**Low noble metal concentrations most effective in gas sensing**

[1] Mädler L, Sahn T, Gurlo A, Grunwaldt JD, Barsan N, Weimar U, SEP. *J Nanopart Res*. 2006;8:783-96,

[2] Suematsu, K.; Shin, Y.; Hua, Z.Q.; Yoshida, K.; Yuasa, M.; Kida, T.; Shimanoe, K., *ACS Appl Mater Inter*, 2014, 6:5319-5326



C=0.5 M, P/D=5/5  
 IICP-OES of leachate  
 and digested solution